# **(Pyridylcyanomethyl)piperazines as Orally Active PAF Antagonists**

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A series of (pyridylcyanomethyl)piperazines was prepared and evaluated for PAF-antagonist activity. Compounds were tested in vitro in a PAF-induced platelet aggregation assay and in vivo in a PAF-induced hypotension test in normotensive rats. Oral activity was ascertained through a PAF-induced mortality test in mice. The main structure-activity trends of the series were established. Activity was mainly found in four skeletons: l-acyl-4-(3-pyridylcyanomethyl) piperazine,l-acyl-4-(4-pyridylcyanomethyl)piperazine,l-acyl-4-(3-pyridylcyanomethyl)piperidine, and l-acyl-4-cyano-4-(3-pyridylamino)piperidine. The acyl substituents, diphenylacetyl and 3,3 diphenylpropionyl, provided the most active compounds, and the introduction of an amine or hydroxy group in the 3,3-diphenylpropionyl substituentled to further improvement in oral activity. As a result, three of the most active compounds (100,114, and 115) have been selected for further pharmacological development.

### Introduction

There has been an ongoing effort to discover PAF antagonists in the last decade. Although PAF has been recognized as having a key role in numerous pathological processes, no clear clinical application of this class of compounds has yet been firmly demonstrated. Medicinal chemists nevertheless continue to shed light on the structure-activity relationships that govern this field by describing new products and optimizing structural variables. The search for novel chemical entities has focused on the preparation of charged structural analogues of PAF, the screening of natural products, and the evaluation of a variety of synthetic compounds, some having known pharmacological activity.<sup>1</sup> Among synthetic compounds, triazolodiazepine is the nucleus that has received the most attention and in which there are orally potent PAF antagonists, such as compound 1 (WEB 2086).<sup>2</sup> Other chemically unrelated structures in this group are pyrrolodihydrothiazole 2 (RP 59227),<sup>3</sup> dihydroimidazoisoquinoline  $3$  (SDZ 64-412),<sup>4</sup> pyridylthiazolidine 4 (YM 461),<sup>5</sup> and pyridocarboxamide 5 (RO 24-0238).<sup>6</sup>

We recently described a series of ionic tetrahydrofuran, dioxolane, and glycol derivatives with potent PAF antagonist activity. $7$  In the present paper we report the results of a parallel search for novel noncharged structures with good oral bioavailability.

From a preliminary screening, bis(3,4,5-trimethoxybenzoyl) piperazine<sup>8</sup> (30), a compound described in 1968 as analgesic, was identified as a relatively potent PAF antagonist in vitro and in vivo. In an effort to prepare more potent compounds, we explored the chemical modification of this structure to identify structural features necessary for optimal activity. We first ascertained the effect of replacing one of the 3,4,5-trimethoxybenzoyl substituents at the symmetric known compound 30 by a pyridine-containing substituent, leading us to identify cyanomethylpyridine 36 as a lead compound. In fact, the pyridine ring and the amide group separated by a given distance found in this molecule is a common feature in several known PAF antagonists (see structures 2, 4, and 5). Subsequent synthetic and structure-activity studies based on 36 resulted in the preparation of a new class of potent orally active PAF antagonists. The present work discusses the synthesis and evaluation of the PAFantagonist activity of these compounds.

### **Chemistry**

The carboxamides 31-36 (Table I) and compounds 66- 69,71,75,79, and 80 (Table IV) were prepared as outlined in Scheme I. Acylation of piperazine with 3,4,5-trimethoxybenzoyl chloride under controlled pH gave a 1.7:1 mixture of amine  $7a$  and compound 30. The N-formyl derivatives  $6b$ , c were obtained by acylation of  $N$ -formylpiperazine with appropriate carboxylic acids using HOBT and DCC in DMF (method A). Acid hydrolysis of 6b,c yielded piperazines 7b,c, which together with 7a were the starting material for a series of reactions. Compounds 31, 32, 68, and 69 were prepared from 7a-c by a second acylation with the appropriate carboxylic acid by method

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*Orally Active PAF Antagonists* 

## **Chart I**



 $^a$  (a) 3,4,5-Trimethoxybenzoyl chloride, THF/H<sub>2</sub>O, pH = 2-3.5, room temperature, 2 h; (b) RCOOH, DCC, HOBT, DMF, room temperature, 18 h (method A); (c) aqueous HCl, MeOH, 100 <sup>0</sup>C, 1 h; (d) pyridine-3-carboxaldehyde, sulfur, DMF, 100 <sup>0</sup>C, 3 h; (e) pyridine-2-carboxaldehyde or pyridine-3-carboxaldehyde, NaBHsCN, MeOH, pH = 6-8, room temperature, 24 h; (f) 3-acetylpyridine, NaBHsCN, MeOH, pH - 6-8, room temperature, 24 h; (g) (1) 3-acetylpyridine, TsOH, toluene, reflux; (2) KCN, AcOH, room temperature, 18 h; (h) 3-(chloromethyl)pyridine hydrochloride, TEA, CHCl3, room temperature, 48 h; (i) pyridine-3-sulfonyl chloride hydrochloride, TEA, CHCl3, room temperature, 18 h; (j) 3-(2-bromoacetyl)pyridine hydrobromide, TEA, CHCl3, room temperature, 18 h; (k) NaBH4, MeOH, 0 °C, 0.5 h; (1) SOCl2, CHCl3, room temperature, 1 h, then reflux, 1 h; (m) KCN, DMSO, 90 °C, 6 h; (n) ArCHO, KCN, MeOH/H<sub>2</sub>O, pH = 7, room temperature, 18 h.



<sup>a</sup> (a) Pyridine-3-carboxaldehyde, KCN, MeOH/H<sub>2</sub>O, pH = 7, room temperature, 18 h; (b) RCOOH, DCC, HOBT, DMF, room temperature, 18 h (method A); (c) RCOOCl, TEA, CHCl<sub>3</sub>, room temperature, 18 h (method B); (d) pyridine-4-carboxaldehyde, KCN, MeOH/H<sub>2</sub>O, pH = r, room temperature, 16 n, (e) reagent prepared from 5-1 $\lambda$ -(uphenymetry), ammofproparote acta,  $\lambda$ -hydroxysuccinimae; DCC, DME, foom<br>temperature, 2 h; DME, room temperature, 2 h, and then 60 °C, 30 min; (f) chloroacety CHCI3, room temperature, 48 h (method C).

A. Treatment of 7a with pyridine-3-carboxaldehyde and sulfur in DMF<sup>9</sup> furnished compound 33. Reaction of 7a with pyridine-3-carboxaldehyde and NaBH<sub>3</sub>CN<sup>10</sup> led to a 1.2:1 mixture of amine 34 and cyanomethylamine 36. Under the same conditions, the reaction of 7b with 3-acetylpyridine gave mainly compound 75. Condensation of 7b with 3-acetylpyridine under acid catalysis, followed by treatment with KCN in HOAc, afforded compound 80.<sup>11</sup> Amines 66 and 67 were prepared by alkylation of **7b,c**  with 3-(chloromethyl)pyridine hydrochloride. Sulfonamide **71** was prepared by coupling 7c with pyridine-3 sulfonyl chloride hydrochloride.<sup>12</sup>

The homologous cyanomethylpyridine 79 was synthesized through the alkylation of **7b** with 3-(2-bromoacetyl) pyridine<sup>13</sup> followed by reduction with sodium borohydride, conversion of the resulting alcohol 9 into chloride 10 by treatment with SOCl<sub>2</sub>, and finally reaction with KCN in DMSO.

Compounds **81-90** listed in Table V were prepared by reacting 7c with the appropriate aromatic or heterocyclic aldehydes in the presence of KCN.<sup>11</sup> The substituted benzaldehydes and l-methylpyrrole-3-carboxaldehyde are commercial products, and the remaining heterocyclic aldehydes were prepared as described in the literature.<sup>14</sup>

Looking for a more efficient way to prepare l-acyl-4- (3-pyridylcyanomethyl)piperazines, we tried direct condensation of pyridine-3-carboxaldehyde with an excess of piperazine in the presence of KCN. This reaction gave an 85% yield of cyanomethylpiperazine **11** accompanied by only traces of dimer product; we were thus able to avoid use of a protecting group for piperazine (see Scheme II). All compounds listed in Table II were prepared by coupling amine 11 with the appropriate carboxylic acid by means of the aforementioned standard method A or through the mediation of an acid chloride (method B).

Under conditions similar to those used for preparation of **11,** intermediates 12,13, and 14 were obtained in 82%, 95% and 24% yields, respectively, starting from the appropriate aldehyde and amine. In coupling 12 with carboxylic acids to give compounds 81, 111, 116, and **107**  by method A, concomitant formation of the corresponding isonicotinoyl piperazine derivative (for example 117 in the case of  $R = R_3$ ) was observed. A related process, the base-catalyzed autoxidation of  $\alpha$ -amino nitriles as an efficient method for converting aldehydes to amides, has been recently reported.<sup>15</sup> Undesired amide formation

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**Scheme III<sup>a</sup>**



**«(a) 11, pyr, reflux, 48 h; (b) 11, K2CO3, DMF, 80 <sup>0</sup>C, 12 h; (c) (1) 3,3-diphenylpropanoyl chloride, TEA, CHCl3, room temperature, 18 h; (2) POCl3, toluene, 60 <sup>0</sup>C, 0.5 h, and then DME, NaBH4, room temperature, 18 h; (d) 10% NaOH, EtOH, 100 <sup>0</sup>C, 18 h; (e) pyridine-**3-carboxaldehyde, KCN, MeOH/H<sub>2</sub>O, pH = 7, room temperature, **24 h; (f) chloroacetyl chloride, TEA, CHCl3, room temperature, 18 h; (g) U, TEA, CHCl3, room temperature, 48 h; (h) 12, TEA, CHCl3, reflux, 48 h.** 

**reaction was reduced in the synthesis of 107 by the isolation**  of the N-hydroxysuccinimide ester of 3-[N-(diphenyl**methyl)amino] propionic acid prior to reaction with 12 in DME.** 

## Scheme IV<sup>a</sup>

**Compounds 91-106 (Table VI), which are glycine or 0-alanine derivatives, were obtained through two methods (see Scheme II). In the first, amine 11 was coupled with the previously prepared amino acids<sup>16</sup> by method A to give compounds 99,100,105,106, and 115 (Table VI). In the second, intermediate 15, prepared by the reaction of amine 11 and chloroacetyl chloride, was quickly treated with the appropriate aralkylamine (method C). In this way, compounds 91-98 and 101-104 were obtained.** 

**Compounds 62-64 (Table III) and 110,112 (Table VI) were prepared as shown in Scheme III. Thus, compounds 62 and 63 having urea and carbamate functions, respectively, were formed by the reaction of 11 with intermediates 16 and 17 (eq A and B). For the preparation of 64, N-(ethoxycarbonyl)piperazine was acylated with 3,3 diphenylpropionic acid, and the amide formed was reduced with POCl3 and NaBH4. 17 The resulting alkylpiperazine 18 was subjected to basic hydrolysis to give 19, which was condensed with pyridine-3-carboxaldehyde in the presence of KCN (eq C). Compound HO with the carbonyl group in a shifted position was prepared via coupling of chloroacetyl chloride with (3,3-diphenylpropyl)amine and treatment of the resulting chloride 20 with amine 11 (eq D). Compound 112 was obtained by alkylation of amine**  12 with N-(2-chloroethyl)-N-diphenylmethylamine 21 (eq **E).** 

**The carba analogues of l-acyl-4-(3-pyridylcyanomethyl) piperazines, compounds 77,78, and 114, were prepared by the sequence shown in Scheme IV. Aldol condensation of iV-(ethoxycarbonyl)-4-piperidone with 3-pyridylacetonitrile in refluxing NaOEt/EtOH gave 22, which was deprotected by TMSI generated in situ<sup>18</sup> to yield amine 23, and this was acylated by standard method A to give compound 77. The strategy used to prepare amine 26 was to alkylate 3-pyridylacetonitrile with tosylate 24 followed by deprotection of the amine group, again using TMSI generated in situ. The conversion of 26 to 78 and 114 was accomplished by standard method A.** 

**(Pyridylamino)piperidines 73,76, and 113 were prepared as shown in Scheme V. Thus, ketone 27, which was derived by acylation of 4-hydroxypiperidine using method A followed by oxidation with CrOs/pyridine, was treated with** 



**° (a) 3-Pyridylacetonitrile, EtONa, EtOH, reflux, 1 h; (b) NaI, TMSCl, CH3CN, 50 °C, 48 h; (c) 3,3-diphenylpropanoic acid, DCC, HOBT, DMF, room temperature, 18 h (method A); (d) (1) EtOCOCl, CHCl3, TEA, room temperature, 18 h; (2) TsCl, TEA, CH2Cl2, room temperature,**  18 h; (e) 3-pyridylacetonitrile, NaH, DMF, 50 °C, 18 h; (f) [N-(diphenylmethyl)amino]acetic acid, method A; (g) [N-(diphenylmethyl)amino]acetic **acid, method A.** 

**Scheme V***"* 



*a* (a) (1) 3,3-diphenylpropanoic acid, DCC, HOBT, DMF, room temperature, 18 h (method A); (2) CrO<sub>3</sub>, pyr, CH<sub>2</sub>Cl<sub>2</sub>, room temperature, 18 h; (b) 3-aminopyridine,  $NaBH<sub>3</sub>CN$ ,  $MeOH$ , pH = 6-8, room temperature, 24 h; (c) 3-aminopyridine, KCN, AcOH, 60 °C, 18 h; (d) di-tert-butyl dicarbonate, 1 N NaOH, THF, room temperature, 2 h; (e) (1)  $HCl_{(g)}/dioxane$ , CHCl<sub>3</sub>, 0 °C, 0.5 h; (2) [N-(diphenylmethyl)amino]acetic acid, method A.

3-aminopyridine and NaBHsCN to furnish 73. Reaction of ketone 27 with pyridine-3-carboxaldehyde in the presence of KCN in HOAc<sup>11</sup> gave compound 76. Under similar conditions,  $N$ -BOC-4-piperidone (28) was converted to 29, which was deprotected by treatment with  $HCl_{(g)}$  in dioxane and coupled with  $[N-(diphenylmethyl)-]$ amino] acetic acid using method A to give compound 113.

One analogue bearing a thioamide group (61) was prepared by treating 53 with Lawesson reagent (Scheme VI). Compound 53 was also converted to amide 72 by the action of  $HCl_{(g)}$  in MeOH. Finally, the oxidation of 66 and 69 with MCPBA yielded 70 and 74, respectively.

### **Results and Discussion**

Compounds were evaluated for PAF-antagonist activity using the in vitro PAF-induced platelet aggregation assay<sup>19</sup> and the in vivo PAF-induced hypotension test in normotensive rats.<sup>20</sup> In the latter, initial intravenous doses of 5 mg/kg of the test compounds were administered before PAF injection. Only when more than 50% inhibition of the hypotension response was achieved were more doses tested in order to obtain the  $ID_{50}$  values. Many compounds

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<sup>a</sup> (a) Lawesson's reagent, toluene, reflux, 18 h; (b) HCl<sub>(g)</sub>/MeOH, room temperature, 18 h; (c) MCPBA, CHCl<sub>3</sub>, room temperature, 1 h; (d) MCPBA,  $CH_2Cl_2$ , 0 °C, 1 h.



Figure 1.

were also tested for oral activity in the PAF-induced mortality test in mice.<sup>21</sup>

Table I shows data for compounds in which one of the 3,4,5-trimethoxybenzoyl groups of known compound 30 was changed by a pyridine-containing substituent. The cyanomethylpyridine 36 and especially its analogues 37 and 38 with enlarged aralkyl chains showed a modest but significant increase in activity that encouraged us to search for even more potent compounds based on their structure.

Tables H-VI list a series of compounds generated by the systematic variation of four main features found for the aforementioned lead compound 36: the aralkyl chain, the carboxamide moiety, the cyanomethyl linker, and the aromatic ring (Figure 1).

We first focused our attention on the aralkyl chain (Table II). Within a homogeneous phenylalkanoyl series (39-42), 3-phenylpropanamide 41 was the most potent compound and nearly equipotent to its trimethoxysubstituted analogue 38. Linear alkanoyl derivatives 44- 47 and aralkanoyl derivatives with condensed aromatic rings 48-51 were generally less active than 3-phenylpropanamide 41. However, the presence of a diphenylmethyl moiety in the acyl substituent (53-55 and 59) increased activity in the aggregation test to about 50 times that of 41. The hypotension test was more selective; 3,3-diphenylpropanamide 55 and 3,3-diphenylpropenamide 59 were

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Table I. Characterization and PAF-Antagonist Activity of 3,4,5-Trimethoxybenzoylpiperazine Derivatives





<sup>a</sup> Concentration required to inhibit PAF-induced maximum aggregation by 50%. Parentheses contain 95% confidence limits. <sup>b</sup> Dose required to reduce the lowering of the arterial blood pressure caused by PAF by 50%. Parentheses contain 95% confidence limits. Cose required to inhibit PAF-induced mortality by 50%. Parentheses contain 95% confidence limits. d Solvents: A = acetonitrile, Cl = chloroform, D = dichloromethane, H = hexane, EA = ethyl acetate.  $\epsilon$  Empirical formula with amount of water of hydration. *f* Analytical results for the indicated elements are within  $\pm 0.4\%$  of the calculated values unless indicated otherwise.

the best compounds  $(ID_{50}$  less than 0.1 mg/kg iv). The activities of the last two compounds surpassed those of reference compound 1 and met our threshold activity criteria in the aggregation and hypotension tests although not in the mortality test.

Attempts to improve activity by modifying the carboxamide unit (Table III) were not successful. Only carbamate 63 maintained activity in the two first screening tests; nevertheless, this compound failed in the oral test. As a result, carboxamide moiety was left untouched in the remaining compounds.

A study of the nature of the linker between piperazine and pyridine rings (Table IV) revealed a significant role for the cyano group in determining activity. Neither the replacement of this group  $(66, 67, 72, 74, 75)$  nor the substitution of the cyanomethyl moiety by a carboxamide or sulfonamide (68-71) was compatible with activity, as can be seen by comparison of these compounds with

3-(cyanomethyl) pyridines 53 and 55. That the cyano group is a key feature of this class of compounds was further corroborated by the fact that other compounds carrying this group retained the high potency observed in 53 and 55. Examples are compound 76, in which the amine and cyano group shifted their positions; 77 with a carboncarbon double bond; and compound 78, the carba analogue of 53. Finally, the insertion of a methylene group between the piperazine ring and the cyanomethyl linker (79) or additional substitution in the carbon atom bearing the cyano group (80) led to significant loss of activity.

Investigation into the nature of the aromatic ring (Table V) revealed that the antagonist effect almost disappeared when the pyridine ring was replaced by substituted phenyl rings or five-membered nitrogen-containing heterocycles. With regard to the position of nitrogen in the pyridine, 4-pyridine 81 retained the overall activity of its 3-pyridine analogue 55. It was half as active in the platelet aggregation

 $\bar{z}$ 

**Table II.** The "Arylalkyl Chain": Characterization and PAF-Antagonist Activity of l-Acyl-4-(3-pyridylcyanomethyl)piperazine Derivatives

# R-N J CN



 $\bar{z}$ 

### Table II. (Continued)



CN

*<sup>a</sup>~f* See footnotes in Table I.

Table III. The "Carboxamide Moiety": Characterization and PAF-Antagonist Activity of Compounds 61-65



 $a-f$  See footnotes in Table I.  $s$  H calcd, 5.52; found, 6.06.

and hypotension tests but performed slightly better in the mortality test. The 2-pyridine derivatives were discarded in an earlier stage of this study due to the lack of activity of compound 35 (Table I).

Once the main structure-activity trends of this series had been established, we chose the aralkyl chain to make a final attempt to further improve oral activity, since this variable had shown greater flexibility. The protection shown by amine 64 (Table III) in the mortality test, exceptional considering the compound's level of activity, suggested the idea of inserting an amino group into the aralkanoyl substituent (Table VI). Before examining the oral activity of the new compounds, we checked the activities in the aggregation and hypotension tests. Comparison of these amino acid derivatives with their carba analogues in Table II revealed absolute coincidence in the substitution pattern required for high level of activity. Pairs of compounds which only differ from one another in the insertion of an amino group (see 92, vs 40,93 vs 41, 98 vs 48, and 100 vs 55) showed similar potency at different ranges in the aggregation and hypotension tests. The diphenylmethyl moiety is again present in the most active compounds (100,106, and 107). Comparison of 101 with 106 suggested that the optimal position for the amino group was next to the diphenylmethyl moiety. The additional substitution of the amino group by a methyl group was detrimental to activity in the four pairs of compounds tested (91 vs 92,94 vs 95, 96 vs 97, and 99 vs 100). Other isomeric patterns different from [ (diphenylmethyl)amino] acetyl or [(diphenylmethyl)amino]propionyl (103-105),

although with similar levels of activity in the aggregation test, were inferior in the hypotension test. Disminished activity was also observed for compounds 108-110 with shifted carbonyl group positions, for compound 111 with two amide functions and 112 with two amine functions.

Regarding the results of amino acid derivatives in the PAF-induced mortality test after oral administration, we observed that the incorporation of the amino group decreased the  $ID_{50}$  values 2- or 3-fold (55 vs 99, 100, 106, and 107). Similar improvement was also observed for 3-hydroxy-3,3-diphenylpropanamides 115 and 116. Furthermore, compounds 113 and 114, bearing a  $[N-(diphe$ nylmethyl)amino]acetate substituent incorporated in the other two cyano-containing skeletons prepared, showed better oral activity than their analogues with no amino group in the side chain (76 and 78). We can therefore conclude that the introduction of a polar group with the ability to establish hydrogen bonding in the aralkyl chain increases bioavailability in this series. Interestingly, 4-pyridines incorporating an aminoacyl or hydroxyacyl substituent (107 and 116) were again nearly as potent as their 3-pyridine analogues (106 and 115).

In conclusion, we report a novel series of potent and orally active PAF antagonists: l-acyl-4-(3-pyridylcyanomethyl)piperazine, 1-acyl-4-(4-pyridylcyanomethyl)piperazine, l-acyl-4-(3-pyridylcyanomethyl)piperidine, and l-acyl-4-cyano-4-(3-pyridylamino)piperidine. The PAF antagonist character of these compounds is highly dependent on the presence of a cyano group and on the nature of the aromatic ring, 3- or 4-pyridine being optimal. The

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Table IV. The "Cyanomethyl Linker": Characterization and PAF-Antagonist Activity of Compounds 66-80



### Table IV. (Continued)



**Ph**  $\leftarrow$  **N**  $\leftarrow$  **N-R** 

 $4$ -/ See footnotes in Table I.  $4$  H calcd, 7.16; found, 6.48.

Table V. The "Aromatic Ring": Characterization and PAF-Antagonist Activity of Compounds 81-90



 $a-f$  See footnotes in Table I.  $s$  N calcd, 12.33; found, 12.94.

### highest levels of activity are found in compounds bearing a  $3,3$ -diphenylpropionyl and especially  $[N-(diphenyl$ methyl)amino] acetyl or 3-hydroxy-3,3-diphenylpropionyl as acyl substituents. The most interesting compounds to emerge from the work are **100, 106,** 107, **114,** and 115. These compounds, that compare quite favorably with the reference compound 1 in the aggregation an hypotension tests, have been selected for further pharmacological development.

### **Experimental Section**

A. Chemistry. Melting points were determined with a Mettler FP 80 central processor melting-point apparatus and are uncorrected. Melting points of organic salts varied, depending on the amount of water in the sample and should be regarded as approximate. Infrared spectra were recorded on a Perkin-Elmer 983 spectrophotometer. <sup>1</sup>H NMR (80 MHz) and <sup>13</sup>C NMR (20.1 MHz) spectra were recorded on a Brücker AC80 spectrometer and are reported in ppm on the *S* scale from the indicated reference. Combustion analyses were performed with a Carlo **Table VI. The Incorporation of an Amino or Hydroxy Group in the "Arylalkyl Chain": Characterization and PAF-Antagonist Activity of Compounds 91-116** 





Table VI. (Continued)



 $4$ <sup>-/</sup> See footnotes in Table I.  $8$  H calcd, 6.20; found, 5.64.  $h$  N calcd, 14.85; found, 13.69.

Erba 1106 analyzer. Liquid chromatography was performed with a forced flow (flash chromatography) of the indicated solvent system on SDS Silica Gel Chromagel 60 a C.C. (230-400 mesh). Analytical thin-layer chromatography (TLC) was performed using Macherey-Nagel 0.25-mm silica gel SIL G-25 plates. When necessary, solvents and reagents were dried prior to use. THF, Et<sub>2</sub>O, and toluene were distilled from sodium metal/benzophenone ketyl. CHCl<sub>3</sub> was passed through an alumina column. CH<sub>2</sub>- $Cl<sub>2</sub>$  and Et<sub>3</sub>N were distilled from CaH<sub>2</sub>. DMSO and DMF were distilled under reduced pressure from  $CaH<sub>2</sub>$  and stored over activated 4-Å molecular sieves. Unless otherwise specified, all nonaqueous reactions were conducted in a rigourously dried argon atmosphere, using oven-dried glassware.

C-18-PAF-acether was synthesized from (S)-batyl alcohol<sup>22</sup> following a published procedure.<sup>23</sup> Compound 1 was kindly provided by Boehringer Ingelheim.

1-(3,4,5-Trimethoxybenzoyl)piperazine (7a). 2 NHCl was added to a solution of piperazine (9.9 g, 0.114 mol) in water (90 mL) and THF (45 mL) until  $pH = 2$  was attained. Then,  $40\%$ aqueous NaOAc and a solution of 3,4,5-trimethoxybenzoyl chloride (0.114 mol) in THF (45 mL) were added alternately so that the pH was kept within a range of  $2-3.5$ . The resultant mixture was stirred at room temperature for 2 h. Then, it was saturated with  $K_2CO_3$  and extracted with CHCl<sub>3</sub>. The organic layer was washed with saturated aqueous NaCl solution, dried over anhydrous  $Na<sub>2</sub>SO<sub>4</sub>$ , and concentrated under reduced pressure to give 37.05 g of crude product. Purification by chromatography on silica gel, eluting with CHCl<sub>3</sub>/MeOH/NH<sub>3</sub> (60:2:0.2), yielded 20 g of compound  $30^8$  and 21.4 g (67%) of 7a as an oil: IR (film) v 3221, 2925, 2827, 1623, 1573, 1404, 1326, 1231 cm<sup>-1</sup>, <sup>1</sup>H NMR (80 MHz, CDCl<sub>3</sub>) δ (TMS) 6.62 (s, 2 H, Ar), 3.87 (s, 9 H, MeO), 3.59 (m, 4 H, pip), 2.88 (m, 4 H, pip), 1.79 (s, 1 H, NH).

1-(Diphenylacetyl)piperazine (7b). Method A. To a cooled (0 °C) mixture of N-formylpiperazine (9 mL, 90 mmol), diphenylacetic acid (18.6 g, 90 mmol), and HOBT (13.5 g, 0.1 mol) in DMF (200 mL), was added DCC (21 g, 0.1 mol). The mixture was stirred at room temperature overnight and diluted with EtOAc (700 mL). The insoluble solid was filtered off and the filtrate washed with saturated NaHCO<sub>3</sub> solution and water. The organic layer was dried over MgSO<sub>4</sub>, filtered, and concentrated under reduced pressure to an oil (6b, 25.05g). To a solution of that crude in MeOH (170 mL) was added 10% HCl (300 mL), and the mixture was heated at 100 °C for 1 h. The organic solvent was removed, and the aqueous solution was extracted with CHCl<sub>3</sub>, basified with 5 N NaOH solution and extracted again with CHCl<sub>3</sub>. This second organic layer was dried over MgSO<sub>4</sub>, filtered and concentrated under reduced pressure to an oil. Purification by chromatography on silica gel, eluting with CHCl<sub>3</sub>/MeOH/NH<sub>3</sub> (60:2:0.2), gave 16.25 g (71%) of a white solid: IR (film)  $\nu$  3340, 3021, 2998, 2913, 1632, 1428, 1223, 1031 cm<sup>-1</sup>; <sup>1</sup>H NMR (80 MHz, CDCl<sub>3</sub>)  $\delta$  ppm (TMS) 7.24 (m, 10 H, Ph), 5.18 (s, 1 H, CHPh<sub>2</sub>), 3.50 (m, 4 H, pip), 2.63 (m, 5 H, pip, NH).

1-(3,3-Diphenylpropionyl)piperazine (7c) was prepared as described above (49%): IR (KBr) v 3333, 3017, 2952, 1612, 1447, 1434, 1359, 1316 cm<sup>-1</sup>;<sup>1</sup>H NMR (80 MHz, CDCl<sub>3</sub>) δ ppm (TMS) 7.23 (m, 10 H, Ph), 4.66 (t,  $J = 7.5$  Hz, 1 H, CHPh<sub>2</sub>), 3.49 (m, 2) H, pip), 3.28 (m, 2 H, pip), 3.03 (d,  $J = 7.5$  Hz, 2 H,  $CH_2CHPh_2$ ), 2.60 (m, 4 H, pip), 1.59 (s, 1 H, NH).

In a similar manner, compounds 31 and 32 (Table I) and 68 and 69 (Table IV) were prepared by reaction of 3-(3,4,5trimethoxyphenyl) propionic acid, nicotinic, and isonicotinic acid with the appropriate piperazine 7a-c.

1-(3,4,5-Trimethoxybenzoyl)-4-thionicotinoylpiperazine (33). A mixture of  $7a$  (1.5 g, 5.4 mmol), pyridine-3carboxaldehyde (0.5 mL, 5.4 mmol), and sulfur (0.204g, 5.4 mmol) in DMF (3 mL) was heated at 100 °C for 3 h. The resulting mixture was poured into water and extracted with EtOAc. The

<sup>(22)</sup> Golding, B.; Ioannou, P. Rapid Synthesis of 3-O-Benzyl-sn-glycerol and 2-O-Benzylglycerol. Synthesis 1977, 423-424.<br>
(23) Wissner, A.; Kohler, C. A.; Golstein, B. M. Analogues of Platelet

Activating Factor 5. Multiple Oxygen Substitution of the Alkoxy Chain.<br>J. Med. Chem. 1986, 29, 1315–1319.

<sup>&</sup>quot; (24) Ohta, S.; Nozaki, A.; Ohashi, N.; Nishimura, Y.; Yoneda, Y.; Okamoto, Masao. An Efficient Biogenetic-Type Transformation of 3-Tertiaryhydroxyalkanoic Acids to Olefinic Compounds. Chem. Pharm. Bull. 1986, 34, 5144-5148.

organic phase was washed with water, dried over anhydrous Na2-SO4, and filtered. The filtrate was concentrated in vacuo to an oil residue  $(2.2 g)$ , which was purified by chromatography (CHCl<sub>3</sub>/  $MeOH/NH<sub>3</sub> 60:2:0.2)$  to afford 1.18 g (55%) of a yellow solid: mp 138-139 <sup>0</sup>C; IR (KBr) *v* 2931,2857,1666,1632,1580,1485,1462, 1421, cm<sup>-1</sup>; <sup>1</sup>H NMR (80 MHz, CDCl<sub>3</sub>) δ ppm (TMS) 8.55 (m, 2 H, pyr), 7.61 (d t, *J* = 7.6 Hz, *J* = 1.8 Hz, 1 H, pyr), 7.31 (dd, 1 *YL, J=* 7.9 Hz, *J* = 4.8 Hz, pyr), 6.66 (s, 2 H, Ar), 3.85 (s, 9 H, MeO), 3.70 (m, 8 H, pip). The product (0.3 g) was dissolved in EtOAc and converted to its hydrochloride by treatment with  $HCl_{(g)}/Et_2O$  solution to give a yellow solid (0.2 g, 63%): mp 231-232 <sup>0</sup>C. Anal. (C2OH23N3SO4-HCl-V2H2O) C, **H,** N.

**l-(3,4,5-Trimethoxybenzoyl)-4-(3-pyridylmethyl)piperazine (34) and l-(3,4,5-Trimethoxybenzoyl)-4-(3-pyridylcyanomethyl)piperazine (36).** To a solution of **7a** (1.5 g, 5.3 mmol) in MeOH (10 mL) and a few drops of 2 N HCl(g)/MeOH solution to adjust pH at 6-8 was added pyridine-3-carboxaldehyde (0.5 mL, 5.33 mmol) and NaBH3CN (0.33 g, 5.3 mmol). The resulting solution was stirred at 25 °C for 24 h. Then, additional NaBH<sub>3</sub>-CN (0.1 g, 1.6 mmol) was added and stirring continued for 48 h. The reaction mixture was then treated with 0.1N NaOH solution and the resulting suspension extracted with EtOAc. The organic phase was dried over  $Na<sub>2</sub>SO<sub>4</sub>$  and fully evaporated. The residue was purified by chromatography on silica gel (CHCl<sub>3</sub>/MeOH/ NH3 60:2:0.2) to yield 0.75 g (38%) of **34** and 0.66 g (31 *%)* of 36 as colorless oils. **34:** IR (film) *v* 2933, 2825, 1625, 1579, 1455,  $1421, 1325 \text{ cm}^{-1}$ ; <sup>1</sup>H NMR (80 MHz, CDCl<sub>3</sub>)  $\delta$  ppm (TMS) 8.51 (m, 2 H, pyr), 7.63 (broad d, *J* = 8.0 Hz, 1 H, pyr), 7.27 (dd, *J*   $= 8.0$  Hz,  $J = 4.8$  Hz, 1 H, pyr), 6.62 (s, 2 H, Ar), 3.85 (s, 9 H,  $CH<sub>3</sub>O$ , 3.62 (m, 4 H, pip), 3.55 (s, 2 H), 2.48 (m, 4 H, pip); <sup>13</sup>C NMR (20.15 MHz, CDCl3) S ppm (TMS) 169.77 (C, CO), 153.09 (C, pyr), 150.15 (CH, pyr), 148.59 (CH, pyr), 139.5 (C), 136.32 (CH, pyr), 132.87 (C), 130.85 (C), 123.11 (CH, pyr), 104.41 (CH, Ar), 60.58 (CH<sub>3</sub>, OCH<sub>3</sub>), 59.76 (CH<sub>2</sub>), 56.07 (2 CH<sub>3</sub>, OCH<sub>3</sub>), 52.77  $(CH_2, pip)$ , 45.00 (CH<sub>2</sub>, pip). The product was dissolved in EtOAc and converted to its hydrochloride by treatment with  $\text{HCl}_{(g)}$ / and converted to its hydrochlonde by treatment with ricity,  $Et<sub>20</sub>$  and  $Et<sub>20</sub>$  and  $Et<sub>20</sub>$  and  $Et<sub>20</sub>$  and  $Et<sub>20</sub>$  and  $Et<sub>20</sub>$  and  $F<sub>20</sub>$  and  $F<sub>20</sub>$  and  $F<sub>20</sub>$  and  $F<sub>20</sub>$  and  $(C_{20}H_{25}N_3O_4.2HCl·2H_2O)$  C, H, N.

36: IR (film)  $\nu$  2935, 2827, 1624, 1580, 1458, 1421, 1327 cm<sup>-1</sup>; <sup>1</sup>H NMR (80 MHz, CDCl<sub>3</sub>) δ ppm (TMS) 8.79 (broad s, 1 H, pyr), 8.65 (broad d, *J* = 3.6 Hz, 1 H, pyr), 7.88 (broad d, *J* = 8.0 Hz, 1 H, pyr), 7.37 (dd,  $J = 8.2$  Hz,  $J = 4.8$  Hz, 1 H, pyr), 6.61 (s, 2) H, Ar), 4.97 (s, 1 H, CHCN), 3.85 (s, 9 H, CH3O), 3.67 (m, 4 H, pip), 2.62 (m, 4 H, pip); <sup>13</sup>C NMR (20.15 MHz, CDCl3) *S* (TMS) 169.86 (C, CO), 153.17 (C, pyr), 150.29 (CH, pyr), 149.11 (CH, pyr), 139.5 (C), 135.20 (CH, pyr), 130.40 (C), 128.22 (C), 123.33  $(CH, pyr)$ , 113.72 (CH, CN), 104.38 (CH, Ar), 60.57 (CH<sub>3</sub>, OCH<sub>3</sub>), 59.68 (CH, CHCN), 56.10 (CH3, OCH3), 49.43 (CH2, pip), 48.00 (CH2, pip). The hydrochloride was prepared as above to give a white solid: mp 86-87 °C. Anal.  $(C_{21}H_{24}N_4O_4.2HC1·H_2O)$  C, H, N.

The procedure was repeated with pyridine-2-carboxaldehyde to give **35** (Table I).

**l-(Diphenylthioacetyl)-4-(3-pyridylcyanomethyl)piperazine (61).** To a solution of **53** (0.238 g, 0.6 mmol) in toluene (10 mL) was added Lawesson reagent (0.24 g, 0.6 mmol). The resulting mixture was heated at reflux for 18 h. Then, the solvents were removed and the residue was purified by chromatography on silica gel (hexane/EtOAc 1:1) to yield 0.18 g (73%) of 61 as a white solid: mp 212-215 <sup>0</sup>C; IR (KBr) *v* 2916,2818,1594,1575, 1474, 1455, 1429, 1412, 1280, 1231 cm<sup>-1</sup>; <sup>1</sup>H NMR (80 MHz, CDCl<sub>3</sub>)  $\delta$  (TMS) 8.66 (m, 2 H, pyr), 7.80 (broad d,  $J = 7.9$  Hz, 1 H, pyr), 7.23 (m, 11H, pyr, Ar), 5.56 (s, 1H, CHCN), 4.84 (s, 1H, CHPh2), 4.49 (m, 2 H, pip), 3.75 (t, *J* = 4.7 Hz, 2 H, pip), 2.69 (t, *J* = 4.7 Hz, 2 H, pip). Anal.  $(C_{25}H_{24}N_4S^{1/2}H_2O)$  C, H, N.

**l-(Diphenylacetyl)-4-(3-pyridylmethyl)piperazine** (66). To a solution of 7b (1.6 g, 10 mmol) and 3-(chloromethyl)pyridine hydrochloride (3 g, 10 mmol) in anhydrous  $CHCl<sub>3</sub>$  (50 mL) was added Et3N (4 mL). The mixture was stirred for 48 h at room temperature. Then, it was washed with water, dried over anhydrous Na<sub>2</sub>SO<sub>4</sub>, and evaporated to dryness. The residue (4.2) g) was purified by chromatography on silica gel  $\rm (CHCl_3/MeOH)$  $3\%$  ) to give 2 g (54%) of 66 as a white solid: mp 149–150 °C; IR (KBr) *v* 3019, 2798, 2761, 1631, 1419, 1295, 1230 cm<sup>-1</sup>; <sup>1</sup>H NMR  $(80 \text{ MHz}, \text{CDCl}_3)$   $\delta$  ppm (TMS) 8.49 (m, 2 H, pyr), 7.60 (broad d, *J* = 7.8 Hz, 1 H, pyr), 7.24 (m, 11 H, pyr, Ph), 5.18 (s, 1 H,  $CHPh<sub>2</sub>$ ), 3.70 (t,  $J = 4.8$  Hz, 2 H, pip), 3.44 (m, 4 H, pip,  $CH<sub>2</sub>$ -pyr), 2.41 (t, *J* = 4.9 Hz, 2 H, pip), 2.14 (t, *J* = 4.6 Hz, 2 H, pip). Anal.  $(C_{24}H_{25}N_3O)$  C, H, N.

The procedure was repeated with **7c** to give 67 (Table IV). 1-(Diphenylacetyl)-4-isonicotinoylpiperazine, N-Oxide (70). To a solution of  $69$  (0.4 g, 1 mmol) in CHCl<sub>3</sub> (2 mL) was added a solution of 50% MCPBA  $(0.320 \text{ g})$  in CHCl<sub>3</sub>  $(1 \text{ mL})$ dropwise. The resulting solution was stirred for 1 h at room temperature and then quenched with 0.5 N NaOH solution. The organic phase was separated and dried over anhydrous Na<sub>2</sub>SO<sub>4</sub>, and the solvent was removed. The residue was purified by chromatography on silica gel (CHCl<sub>3</sub>/MeOH 5%) to afford 0.27 g (77 %) of 70 as a colorless oil that crystallized upon addition OfEt2O: mp 182-183 <sup>0</sup>C; IR (KBr) *v* 3075,1623,1489,1448,1429, 1249 cm<sup>-1</sup>; <sup>1</sup>H NMR (80 MHz, CDCl<sub>3</sub>) δ (TMS) 8.22 (d,  $J = 6.8$ ) Hz, 2 H, pyr), 7.25 (m, 12 H, pyr, Ph), 4.67 (t, 1 H, *J* = 7.5 Hz, CHPh<sub>2</sub>), 3.35 (m, 8 H, pip), 3.08 (d,  $J = 7.6$  Hz, 2 H, CH<sub>2</sub>CHPh<sub>2</sub>). Anal. (C25H25N3O3-V4H2O) C, **H,** N.

**l-(3,3-Diphenylpropionyl)-4-(3-pyridylsulfonyl)piperazine (71).** A solution of pyridine-3-sulfonyl chloride hydrochlo- $\text{ride}^{12}(0.3 \text{ g}, 1.7 \text{ mmol})$  in  $\text{CHCl}_3$  (15 mL) and  $\text{Et}_3\text{N}$  (0.7 mL) was stirred 5 min, and to that solution was added **7c** (0.5 g, 1.7 mmol) and the stirring continued for 18 h. Then the reaction mixture was quenched with water (20 mL). The organic phase was separated, dried over anhydrous Na<sub>2</sub>SO<sub>4</sub>, and filtered, and the filtrate was concentrated to a brown solid (0.7 g). Purification by chromatography (CHCl<sub>3</sub>/MeOH,  $3\%$ ) afforded 0.38 g (49%) of 71 as a white solid: mp 198-199 <sup>0</sup>C; IR (KBr) *v.* 3020, 2832, 1612, 1487, 1464, 1419, 1352, 1324, 1175 cm<sup>-1</sup>; <sup>1</sup>H NMR (80 MHz, CDCl3) *b* ppm (TMS) 8.89 (m, 2 H, pyr), 7.97 (d, of t, *J* = 8.0 Hz, *J* = 1.8 Hz, 1H, pyr), 7.51 (dd, *J* = 4.7 Hz, *J* = 7.8 Hz, 1H, pyr), 7.13 (m, 10 H, Ph), 4.54 (t,  $J = 7.7$  Hz, 1 H, CHPh<sub>2</sub>), 3.58 (m, 2) H, pip), 3.40 (m, 2 H, pip), 2.97 (d,  $J = 7.6$  Hz, 2 H,  $CH_2CHPh_2$ ),  $2.90 \text{ (m, 2 H, pip)}$ ,  $2.51 \text{ (m, 2 H, pip)}$ . Anal.  $(C_{24}H_{25}N_3O_3S^{5/2})$ **<sup>4</sup>H20)** C, H, N.

**l-(Diphenylacetyl)-4-[l-(aminocarbonyl)-l-(3-pyridyl) methyl]piperazine** (72). Through a cooled (0 "C) suspension of  $53(0.5 g, 1.2 mmol)$  in MeOH  $(10 mL)$  was bubbled HCl<sub>(e)</sub> until the solution was saturated. The resulting solution was stirred at room temperature for 18 h. Solvents were removed, and the residue was treated with  $5\%$  aqueous solution of NaHCO<sub>3</sub> (15) mL). The white solid formed was separated and purified by chromatography on silica gel  $\left( \text{CHCl}_3/\text{MeOH}/\text{NH}_3 \right)$  60:2:0.2) to give  $0.5$  g (96%) of 72. Crystallization from  $CH_2Cl_2/Et_2O$  provided an analytical sample: mp 216-217 <sup>0</sup>C; IR (KBr) *v* 3307, 3055, 2999,1680,1632,1423,1225 cm"<sup>1</sup> ; <sup>1</sup>H NMR (80 MHz, CDCl3) *S*  (TMS) 8.49 (m, 2 H, pyr), 7.60 (d of t,  $J = 7.6$  Hz,  $J = 1.8$  Hz, 1 H, pyr), 7.25 (m, 11 H, pyr, Ph), 6.90 (m, 1 H, NH), 6.24 (m, 1 H, NH), 5.12 (s, 1 H, CHPh2), 3.83 (s, 1 H, CHCO), 3.65 (m, 2 H, pip), 3.41 (m, 2 H, pip), 2.37 (m, 2 H, pip), 2.16 (m, 2 H, pip); <sup>13</sup>C NMR (20.15 MHz, CHCl<sub>3</sub>) δ (TMS) 172.51 (C), 179.23 (C), 150.02 (CH), 149.44 (CH), 139.03 (C), 136.34 (CH), 128.81 (CH), 128.41 (CH), 126.94 (CH), 72.37 (CH), 54.62 (CH), 50.69 (CH2), 45.75 (CH<sub>2</sub>), 42.07 (CH<sub>2</sub>). Anal. (C<sub>25</sub>H<sub>26</sub>N<sub>4</sub>O<sub>2</sub>) C, N, H.

**l-(Diphenylacetyl)-4-(3-pyridylmethyl)piperazine4-Oxide (74).** To a cooled (0 °C) solution of 66 (0.5 g, 1.3 mmol) in  $CH<sub>2</sub>Cl<sub>2</sub>$  (10 mL) was added a solution of 50% MCPBA (0.460 g, 1.3 mmol) in  $CH_2Cl_2$  (8 mL) dropwise. The resulting solution was stirred for 1 h at 0 °C and then quenched with 0.5 N NaOH solution. The organic phase was separated and dried over anhydrous Na2SO4, and the solvent was removed. The residue was purified by chromatography on silica gel (CHCl<sub>3</sub>/MeOH 30 $\%$ ) to afford 0.170 g (36%) of **74** as a white solid: mp 50-55 <sup>0</sup>C; IR (KBr) *v* 3600-2950, 1636, 1593, 1425 cm"<sup>1</sup> ; <sup>1</sup>H NMR (80 MHz, CDCl3) *5* (TMS) 8.59 (m, 2 H, pyr), 7.88 (broad d, *J* = 7.5 Hz, 1 H, pyr), 7.25 (m, 11 H, pyr, Ph), 5.19 (s, 1 H, CHPh<sub>2</sub>), 4.7-3.6 (complex signal, 6 H), 3.05 (m, 2 H, pip), 2.67 (m, 2 H, pip); <sup>13</sup>C NMR (20.15 MHz, CDCl<sub>3</sub>) δ (TMS) 170.08 (C), 152.22 (CH), 150.54 (CH), 140.27 (CH), 138.52 (C), 128.59 (CH), 127.17 (CH), 126.96 (CH), 124.79 (C), 123.25 (CH), 72.17 (CH<sub>2</sub>), 63.18 (CH<sub>2</sub>), 62.73 (CH<sub>2</sub>), 54.45 (CH), 40.23 (CH<sub>2</sub>), 36.37 (CH<sub>2</sub>). Anal. (C<sub>24</sub>H<sub>25</sub>N<sub>3</sub>O<sub>2</sub>·<sup>7</sup>/<sub>2</sub>H<sub>2</sub>O) C, H, N.

**l-(Diphenylacetyl)-4-[l-(3-pyridyl)ethyl]piperazine(75)**  was prepared as described above for **34.** Thus, reaction of **7b** (1 g, 3.57 mmol) with 3-acetylpyridine (0.4 mL, 3.57 mmol) and NaBH3CN (0.44 g, 7.14 mmol) yielded 0.22 g (16%) of **75** as a

white solid: mp 136-139 °C; <sup>1</sup>H NMR (80 MHz, CDCl<sub>3</sub>)  $\delta$  ppm (TMS) 8.49 (m, 2 H, pyr), 7.60 (d of t,  $J = 7.8$  Hz,  $J = 1.8$  Hz, 1 H, pyr), 7.23 (m, 11 H, pyr, Ph), 5.15 (s, 1 H, CHPh<sub>2</sub>), 3.68 (t,  $J = 4.5$  Hz, 2 H, pip), 3.42 (t,  $J = 4.5$  Hz, 2 H, pip), 3.38 (q,  $J$  $= 6.7$  Hz, 1 H, CHCH<sub>3</sub>), 2.42 (m, 2 H, pip), 2.18 (m, 2 H, pip), 1.31 (d,  $J = 6.7$  Hz, 3 H, CHCH<sub>3</sub>). Anal. (C<sub>25</sub>H<sub>27</sub>N<sub>3</sub>O-<sup>1</sup>/<sub>4</sub>H<sub>2</sub>O)C, H, N.

**l-(Diphenylacetyl)-4-[2-(3-pyridyl)-2-oxoethyl]piperidine** (8). To a solution of 7b  $(4.44 \text{ g}, 15 \text{ mmol})$  in CHCl<sub>3</sub>  $(50 \text{ mL})$ was added dropwise a solution previously prepared by addition of  $Et_3N$  (2.36 mL) to a suspension of 3-(2-bromoacetyl)pyridine hydrobromide<sup>13</sup> (4.23 g, 1.5 mmol) in CHCl<sub>3</sub> (80 mL). The mixture was stirred at room temperature for 18 h. It was then washed with 1N NaOH solution, dried, and evaporated to dryness. The residue (8.6 g) was purified by chromatography on silica gel  $(CHCl<sub>3</sub>/MeOH, 3\%)$  to yield 4.56 g (73%) of 8 as a colorless oil: IR (film)  $\nu$  2914, 1692, 1637, 1580, 1427, 1220 cm<sup>-1</sup>; <sup>1</sup>H NMR (80 MHz, CDCl<sub>3</sub>)  $\delta$  (TMS) 9.75 (s, 1 H, pyr), 9.30 (d,  $J = 4.5$  Hz, 1 H, pyr),  $8.78$  (d,  $J = 7.5$  Hz, 1 H, pyr),  $7.79$  (m, 11 H, pyr, Ar), 5.74 (s, 1 H, CHPh<sub>2</sub>), 4.24 (m, 4 H, CH<sub>2</sub>CO, pip), 4.04 (m, 2 H, pip), 3.10 (m, 2 **H,** pip), 2.83 (m, 2 **H,** pip).

**l-(Diphenylacetyl)-4-[2-(3-pyridyl)-2-hydroxyethyl]piperazine (9).** To a solution of ketone 8 (4.56 g, 11 mmol) in MeOH (20 mL), cooled to 0 °C, was added NaBH<sub>4</sub> (0.64 g, 16.5 mmol), and the resulting mixture was stirred at  $0^{\circ}$ C for 30 min. Then 0.1 N HCl solution (10 mL) was added. The mixture was concentrated in vacuo and partitioned between water and CHCl3. The organic phase was separated and dried. The residue was purified by chromatography on silica gel (CHCl3/MeOH, 5 *%)* to yield 2.7 g (62%) of a white solid: mp 134-135 <sup>0</sup>C; IR (KBr) *v*  3389, 2811, 1638, 1448, 1220 cm<sup>-1</sup>; <sup>1</sup>H NMR (80 MHz, CDCl<sub>3</sub>)  $\delta$ (TMS) 9.20 (m, 2 H, pyr), 8.52 (broad d, *J* = 7.6 Hz, 1 H, pyr), 7.99 (m, 11 H, pyr, Ar), 5.93 (s, 1 H, CHPh<sub>2</sub>), 5.47 (t,  $J = 6.4$  Hz, 1 H, CHOH), 4.47 (m, 2 H, pip), 4.23 (m, 2 H, pip), 3.19 (d, J = 6.5 Hz, 2 H, CH<sub>2</sub>CHOH), 3.15 (m, 4 H, pip). Anal.  $(C_{25}H_{27}N_3O_2)$ C, H, N.

**l-(Diphenylacetyl)-4-[2-(3-pyridyl)-2-cyanoethyl]piperazine (79).** To a cooled  $(0 °C)$  solution of alcohol 9  $(1.6 g, 4$ mmol) in CHCl<sub>3</sub> (8 mL) was added  $SOL<sub>2</sub> (0.32 mL)$  in CHCl<sub>3</sub> (4 mL). The mixture was stirred 1 h at room temperature and 1 h at reflux. The solvents were evaporated, and the residue was diluted with 20 mL of CHCl<sub>3</sub>. Then  $Et_3N(1.7 mL)$  was added, and the solution was washed with water. The organic phase was separated and dried over anhydrous Na<sub>2</sub>SO<sub>4</sub>, and the solvent was removed to afford crude chloride 10, which was dissolved in DMSO (10 mL). To this solution was added KCN (0.58 g, 9 mmol). The mixture was heated at 90 °C for 6 h, the solvents were evaporated in vacuo, and the residue was partitioned between CHCl<sub>3</sub> and 0.1 N NaOH solution. The organic phase was dried and evaporated to dryness. The residue was purified by chromatography on silica gel (CHCl3/MeOH, 5%) to yield 0.15 g (10%) of 79 as an oil that crystallized upon addition of Et2O: mp 60-64 <sup>0</sup>C; <sup>1</sup>H NMR (CDCl3 + DMSO-d6) *&* (TMS) 9.18  $(m, 1 H, pyr), 8.81$   $(m, 2 H, pyr), 7.94$   $(dd, J = 8.0$  Hz,  $J = 5.5$ Hz, 1 H, pyr), 7.22 (m, 10 H, Ph), 5.20 (m, 1 H, CHPh<sub>2</sub>), 4.71 (t,  $J = 6.9$  Hz, 1 H, CHCN), 3.85 (m, 2 H, pip), 3.67 (m, 2 H, pip), 3.45 (d, J = 7.0 Hz, 2 H, CH<sub>2</sub>CHCN), 2.82 (m, 2 H, pip), 2.56 (m, 2 H, pip); <sup>13</sup>C NMR (20.15 MHz, CDCl<sub>3</sub>) δ (TMS) 170.03 (C), 161.00 (CH), 149.87 (CH), 149.37 (CH), 139.05 (C), 139.02 (C), 135.20 (CH), 133.05 (C), 129.02 (CH), 128.70 (CH), 128.56 (CH), 127.23 (CH), 127.06 (CH), 123.65 (CH), 117.05 (C), 62.79 (CH), 54.69 (CH), 49.76 (CH<sub>2</sub>), 45.91 (CH<sub>2</sub>), 42.09 (CH<sub>2</sub>), 21.68 (CH<sub>2</sub>). Anal.  $(C_{26}H_{26}N_4O^{.1/2}H_2O)$  C, H, N.

**l-(Diphenylacetyl)-4-[l-(3-pyridyl)-l-cyanoethyl]piperazine** (80). A mixture of amine 7b (1.4 g, 5 mmol), 3-acetylpyridine (0.5 mL, 5 mmol), and  $p$ -TsOH (0.15 g) in toluene (75 mL) was heated in a Dean-Stark for 48 h. The solvent was removed, and the residue was partitioned between CHCl<sub>3</sub> and 0.1 N NaOH solution. The organic phase was separated and dried over Na<sub>2</sub>-SO4, and the solvent was removed to yield 2.5 g of crude product. To a solution of this crude product in glacial HOAc (7 mL) was added KCN (0.5 g, 7.6 mmol). The mixture was stirred at room temperature for 12 h. The solvent was removed in vacuo and the residue worked up with CHCl<sub>3</sub> and 1 N NaOH solution. The organic phase was dried over anhydrous Na<sub>2</sub>SO<sub>4</sub> and the solvent evaporated to dryness. Purification by chromatography of the

residue (1.5 g) on silica gel (hexane/EtOAc, 80%) yielded 0.33 g  $(16\%)$  of a colorless oil that solidified upon addition of Et<sub>2</sub>O: mp 71-76 <sup>0</sup>C; IR **(KBr)** *v* 3055, 3022, 2242, 1642, 1595, 1428, 1414 cm<sup>-1</sup>; <sup>1</sup>H NMR (80 MHz, CDCl<sub>3</sub>) δ (TMS) 8.83 (m, 1 H, pyr), 8.60  $(m, 1 H, pyr)$ , 7.88 (broad d,  $J = 8.2$  Hz, 1 H, pyr), 7.26  $(m, 11$ H, pyr, Ar), 5.19 (s, 1 H, CHPh2), 3.70 (m, 2 H, pip), 3.47 (m, 2 H, pip), 2.48 (m, 4 H, pip), 1.72 (s, 3 H,  $CHCH<sub>3</sub>$ ). Anal.  $(C_{26}H_{26}N_4O^{-1}/_2H_2O)$  C, H, N.

**l-Piperazinyl-3-pyridylacetonitrile (11).** To a solution of piperazine (13.95 g, 0.162 mol) and KCN (5.4 g, 0.083 mol) in water (60 mL) and 1 M,  $pH = 7.1$  phosphate buffer solution (60) mL) was added a solution of pyridine-3-carboxaldehyde (5.1 mL, 0.054 mol) in MeOH (60 mL) dropwise. The solution was stirred 18 h at room temperature, water (80 mL) was added, and the solution was extracted with EtOAc. The organic phase was dried over  $Na<sub>2</sub>SO<sub>4</sub>$  and evaporated to dryness. The residue (17.4 g) was purified by chromatography on silica gel (CHCl3/MeOH/ NH3 60:2:0.2) to yield 8.5 g (81%) of a colorless oil: IR (film) *v*  3301, 2943, 2911, 2225,1588,1573,1474,1449,1440,1419,1123 cm<sup>-1</sup>; <sup>1</sup>H NMR (80 MHz, CDCl<sub>3</sub>)  $\delta$  (TMS) 8.77 (d,  $J = 2.3$  Hz, 1 H, pyr), 8.59 (dd,  $J = 4.8$  Hz,  $J = 1.4$  Hz, 1 H, pyr) 7.87 (broad d,  $J = 6.3$  Hz, 1 H, pyr), 7.36 (dd,  $J = 6.3$  Hz,  $J = 4.1$  Hz, 1 H, pyr), 4.91 (s, 1 H, CHCN), 2.90 (m, 4 H, pip), 2.55 (m, 4 H, pip), 1.82 (s, 1 H, NH); <sup>13</sup>C NMR (20.15 MHz, CDCl<sub>3</sub>)  $\delta$  (TMS) 149.41 (CH), 148.82 (CH), 134.83 (CH), 128.38 (C), 122.76 (CH), 113.83 (C), 59.81 (CH), 50.29 (CH<sub>2</sub>), 45.01 (CH<sub>2</sub>).

**l-Piperazinyl-4-pyridylacetonitrile (12)** was prepared as described above for 11, starting from pyridine-4-carboxaldehyde. After purification by chromatography on silica gel  $(CHCl<sub>3</sub>/MeOH/$ NH3 60:2:0.2) a yellow oil was obtained (82% yield): IR (film) *v* 3296, 2943, 2822, 1594, 1410, 1322 cm<sup>-1</sup>; <sup>1</sup>H NMR (80 MHz, CDCl<sub>3</sub>)</sub>  $\delta$  (TMS) 8.68 (d,  $J = 6.2$  Hz, 2 H, pyr), 7.50 (d,  $J = 6.2$ Hz, 2 H, pyr), 4.81 (s, 1 H, CHCN), 2.96 (t,  $J = 4.8$  Hz, 4 H, pip), 2.55 (t, *J* = 4.8 Hz, 4 **H,** pip), 1.78 (s, 1 H, NH).

**l-(2-Aminoethyl)-4-(3-pyridylcyanomethyl)piperazine(14)**  was prepared as described above for 11, starting from l-(2 aminoethyl)piperazine (24% yield): IR (film) *v* 3361,2941,2880, 1527, 1573, 1451, 1420 cm<sup>-1</sup>; <sup>1</sup>H NMR (80 MHz, CDCl<sub>3</sub>) δ (TMS) 8.76 (d,  $J = 2.4$  Hz, 1 H, pyr), 8.62 (dd,  $J = 1.6$  Hz,  $J = 4.8$  Hz, 1 H, pyr), 7.87 (d of t,  $J = 1.6$  Hz,  $J = 8.0$  Hz, 1 H, pyr), 7.38 (dd, *J* = 5.0 Hz, *J =* 7.6 Hz, 1 H, pyr), 4.96 (s, 1 H, CHCN), 2.57 (m, 12 H, NH2CH2CH2, pip), 1.75 (m, 2 **H,** NH2).

**l-(3,3-Diphenylpropionyl)-4-(3-pyridylcyanomethyl)pip**erazine (55). Method B. To a cooled (0 °C) solution of 11 (1) g, 4.9 mmol) and  $Et_3N$  (1.2 mL) in  $CHCI_3$  (20 mL) was added a solution of 3,3-diphenylpropionyl chloride (1.37 g, 5.6 mmol) in CHCl3 (5 mL) dropwise. The mixture was stirred 18 h at room temperature. The organic solution was washed with water, dried  $(Na<sub>2</sub>SO<sub>4</sub>)$ , and evaporated. The residue (2.4 g) was purified by chromatography on silica gel (EtOAc) to yield 1.28 g (53%) of an oil that crystallized upon addition of Et<sub>2</sub>O: mp 160-161 °C; IR (KBr) *v* 3049, 3025, 2933,1612,1476,1454,1421,1323,1258 cm<sup>-1</sup>;<sup>1</sup>H NMR (80 MHz, CDCl<sub>3</sub>) δ (TMS) 8.67 (m, 2 H, pyr), 7.82  $(broad d, J = 7.9 Hz, 1 H, pyr), 7.23 (m, 11 H, pyr, Ar), 4.82 (s,$ 1 H, CHCN), 4.64 (t,  $J = 7.6$  Hz, 1 H, CHPh<sub>2</sub>), 3.55 (m, 2 H, pip), 3.35 (m, 2 H, pip), 3.04 (d  $J = 7.5$  Hz, 2 H, CH<sub>2</sub>CO), 2.32 (m, 4 H, pip). Anal.  $(C_{26}H_{26}N_4O)$  C, H, N.

The procedure was repeated with appropriate acid chlorides to yield compounds 39-41 and 43-48 (Table II). Compound 65 (Table IV) was also obtained by the same methodology from alcohol **13.** 

**l-(3,3-Diphenyl-3-hydroxypropionyl)-4-(3-pyridylcyanomethyl)piperazine (115). Method A.** To a cooled (0 °C) mixture of 3,3-diphenyl-3-hydroxypropionic acid<sup>24</sup> (1.7 g, 7 mmol), compound 11 (1.2 g, 6 mmol), and HOBT (0.8 g, 6 mmol) in DMF (15 mL) was added DCC (1.22 g, 6 mmol). The mixture was stirred 18 h at room temperature. The solvents were then removed in vacuo, and the residue was stirred with EtOAc. The white solid formed was removed, and the organic solution was washed with saturated NaHCO<sub>3</sub> solution, water, and brine, dried  $(Na<sub>2</sub>SO<sub>4</sub>)$ , and evaporated to dryness. The residue  $(2.8 g)$  was purified by chromatography (EtOAc) to yield 1.62 g (61%) of a white solid: mp 68-74 <sup>0</sup>C; IR (KBr) *v* 3345, 3051, 2914, 1612, 1445, 1417, 1230 cm<sup>-1</sup>; <sup>1</sup>H NMR (80 MHz, CDCl<sub>3</sub>) δ (TMS) 8.69 (m, 2 H, pyr), 7.85 (broad d,  $J = 8.5$  Hz, 1 H, pyr), 7.31 (m, 11 H, pyr, Ar), 4.87 (s, 1 H, CHCN), 3.53 (m, 4 H, pyr), 3.20 (s, 2

**H**, CH<sub>2</sub>CO), 2.45 (t,  $J = 5.0$  Hz, 4 H, pyr). Anal.  $(C_{26}H_{26}N_4O_2)$ **V2H2O) C, H, N.** 

**l-[[.V-(Diphenylmethyl)amino]acetyl]-4-(3-pyridylcyanomethyl)piperazine (100) was prepared as above starting from [N-(diphenylmethyl)aminojacetic acid.<sup>16</sup> After purification by**  chromatography on silica gel (CHCl<sub>3</sub>/MeOH/NH<sub>3</sub> 60:2:0.2), a **colorless oil was obtained (73% yield): IR (film)** *v* **3315, 3052, 3019, 2911,1643,1419,1235, 748, 706 cm'<sup>1</sup> ; <sup>1</sup>H NMR (80 MHz, CDCl3) « (TMS) 8.75 (m, 1 H, pyr), 8.64 (broad d,** *J* **= 5 Hz, 1 H, pyr), 7.83 (d,** *J* **= 7.6 Hz, 1 H, pyr), 7.30 (m, 11 H), 4.88 (s, 1 H), 4.84 (s, 1 H), 3.65 (m, 2 H, pip), 3.36 (s, 2 H, CH2CO), 3.30 (m, 2 H, pip), 2.48 (m, 5 H, pip, NH). The product was dissolved in CH2Cl2 and converted to its hydrochloride by treatment with**   $\text{HCl}_{\text{(g)}}/\text{Et}_2\text{O}$  solution to give a white solid: mp 165-168 °C; IR **(KBr)** *v* **3600-2400,1649,1543,1449,1427,1245,1000 cm"<sup>1</sup> ; <sup>1</sup>H NMR (80 MHz, CDCl3)** *S* **(TMS) 8.87 (m, 3 H, pyr), 8.14 (dd,** *J*  **= 7.6 Hz,** *J* **= 5 Hz, 1 H, pyr), 7.52 (s, 10 H), 5.64 (s, 1 H), 5.58 (s, 1 H), 4.10 (s, 2 H, CH2CO), 3.59 (m, 2 H, pip), 3.38 (m, 2 H<sup>1</sup> pip), 2.60 (m, 4 H, pip). Anal.** (C<sub>26</sub>H<sub>27</sub>N<sub>5</sub>O-3HCl-2H<sub>2</sub>O) C, H, N.

**The procedure was repeated with appropriate carboxylic acids to yield compounds 37,38,42,49-54 and 56-60 (Table II), and 99,105, and 106 (Table VI). In a similar manner, compounds 81 (Table V), 111, and 116 (Table VI) were obtained from amine 12 and compounds 108 and 109 (Table VI) from amine 14.** 

**l-(2-Chloroacetyl)-4-(3-pyridylcyanomethyl)piperazine (15). To a cooled (0 <sup>0</sup>C) solution of 11 (2.5 g, 12.3 mmol) and Et3N (1.25 mL, 12.3 mmol) in CHCl3 (38 mL) was added chloroacetyl chloride (1.0 mL, 12.3 mmol) dropwise. The mixture was stirred 1 h at 0<sup>0</sup>C. After dilution with CHCl3, the organic phase was washed twice with water, dried (Na2SO4), and evaporated. The residue, 3.32 g (97%) of a colorless oil, was immediately used in the following reaction: <sup>1</sup>H NMR (CDCl3)**  *S* **(TMS) 8.79 (d,** *J -* **2.3 Hz, 1H, pyr), 8.67 (broad d,** *J* **= 4.1 Hz, I H, pyr), 7.87 (broad d,** *J* **- 6.3 Hz, 1 H, pyr), 7.39 (dd,** *J* **= 6.3 Hz,** *J* **= 4.1 Hz, 1 H, pyr), 4.97 (s, 1 H, CHCN), 4.08 (s, 2 H, CH2Cl), 3.61 (m, 4 H, pip), 2.64 (m, 4 H, pip).** 

**l-[[AT-(2,2-Diphenylethyl)amino]acetyl]-4-(3-pyridylcyanomethyl)piperazine (101). Method C. To a cooled (0 °C) solution of chloride 15 (1 g, 3.6 mmol) and Et3N (0.5 mL, 3.6 mmol) in CHCl3 (50 mL) was added 2,2-diphenylethylamine (1.42 g, 7.2 mmol). The mixture was stirred 18 h at room temperature. It was then washed with water and the organic layer dried (Na2- SO4) and evaporated to dryness. The residue (2.8 g) was purified**  by chromatography on silica gel (CHCl<sub>3</sub>/MeOH/NH<sub>3</sub> 60:2:0.2) to **yield 1.02 g (66%) of 101 as an oH: <sup>1</sup>H NMR (80 MHz, CDCl3)**  *S* **(TMS) 8.77 (d,** *J* **= 2.5 Hz, 1 H, pyr), 8.64 (dd,** *J* **= 2.5 Hz,** *J*  **• 5 Hz, 1 H, pyr), 7.83 (broad d,** *J* **= 7.6 Hz, 1 H, pyr), 7.30 (m, II H, pyr, Ph), 4.88 (s, 1 H, CHCN), 4.17 (t,** *J* **» 7 Hz, 2 H, CH2NH), 3.41 (m, 7 H), 2.52 (m, 4 H, pip), 1.72 (s, 1H, NH). The product was dissolved in CH2Cl2 and converted to its hydro**chloride by treatment with  $HCl_{(g)}/Et_2O$  solution to give a white **solid: mp 188-192 <sup>0</sup>C; IR (KBr)** *v* **3600-2400,1649,1487,1448, 1248,1000 cm-<sup>1</sup> ; <sup>1</sup>H NMR (80 MHz, D2O)** *S* **(TMS) 8.88 (m, 3 H, pyr), 8.17 (dd,** *J* **- 7.6 Hz,** *J* **- 5 Hz, 1H pyr), 7.42 (m, 10 H, Ph)<sup>1</sup> 5.60 (s, 1H, CHCN), 4.50 (m, 2 H), 4.13 (s, 2 H, CH2CO), 3.9-3.3 (m, 5 H), 2.67 (m, 4 H). Anal. (C27H29N6O-SHCl-H2O) C, H, N.** 

**The procedure was repeated with appropriate aralkylamines to yield compounds 91-98 and 102-104 (Table VI).** 

**l-[3-[JV-(Diphenylmethyl)amino]propionyl]-4-(4-pyridyIcyanomethyl)piperazine (107). A. 3-[N-(diphenylmethyl) amino] propionic acid<sup>15</sup> (5.1 g, 20 mmol) was coupled with 12 (4.3 g, 21 mmol) using method A. There were obtained 0.18 g (15%) of 107 (see below) and 0.28 g (25%) of 117. 117: mp 121-122 <sup>0</sup>C; IR (KBr)** *v* **3478, 3311, 2999, 2855,1632,1593,1487,1427,1283 cm'<sup>1</sup> ; <sup>1</sup>H NMR (80 MHz, CDCl3)** *S* **(TMS) 8.70 (broad d,** *J* **= 6.4 Hz, 2 H, pyr), 7.35 (m, 12 H, pyr, Ar), 4.84 (s, 1 H, CHNH), 3.52**   $(m, 8 H, \text{ pip})$  2.83 (t,  $J = 5.6 Hz$ , 2 H, CH<sub>2</sub>CO), 2.56 (t,  $J = 5.6$ )  $\textbf{Hz, 2 H, CH_2N}, 2.29 \text{ (s, 1 H, NH)}. \textbf{Andl. } (C_{26}H_{28}N_4O_2^{-1}/2H_2O)$ **C, H, N.** 

**B.** To a cooled (0 °C) solution of 3-[N-(diphenylmethyl)amino] propionic acid<sup>15</sup> (5.1 g, 20 mmol) and N-hydroxysuccin**imide (2.3 g, 20 mmol) in 120 mL anhydrous DME was added DCC (4.1 g, 20 mmol). The mixture was stirred at 4<sup>0</sup>C for 18 h. The precipitate was removed, and 12 (4.3 g, 21 mmol) in DME (60 mL) was added to the resulting solution. It was then stirred** 

**2 h at room temperature and 0.5 h at 60 <sup>0</sup>C. The solvents were removed, and the residue was partitioned between EtOAc and saturated NaHCO3 solution. The organic solution was separated and dried (Na2SO4), and the solvents were removed. The residue (10 g) was purified by chromatography on silica gel (CHCl3/ MeOH/NHs 60:2:0.2) to give 3.67 g (46%) of 107 as a colorless oil. A portion was recrystallized from CH2CyEt2O to furnish a white solid: mp 146-148 <sup>0</sup>C; IR (KBr)** *v* **3313, 3020,2908,1636, 1592,1447 cm"<sup>1</sup> ; <sup>1</sup>H NMR (80 MHz, CDCl3)** *6* **(TMS) 8.69 (broad d,** *J* **= 6.4 Hz, 2 H, pyr), 7.35 (m, 12 H, pyr, Ar), 4.86 (s, 2 H, CHCN, CHNH), 3.51 (m, 4 H, pip) 2.88 (t,** *J* **= 5.6 Hz, 2 H,**   $CH_2CO$ ), 2.54 (m, 7 H, pip, NH,  $CH_2N$ ). Anal.  $(C_{27}H_{29}N_6O$ **V3Et2O-V4H2O) C, H, N.** 

**l-[[N-(Diphenylmethyl)amino]carbonyl]-4-(3-pyridylcyanomethyl)piperazine (62). To a solution of amine 11 (0.67 g, 3.3 mmol) in pyridine (30 mL) was added 16 (1 g, 3.3 mmol). The mixture was heated at reflux for 2 days. After cooling, it was poured into water (200 mL), extracted with CH2Cl2, and dried over anhydrous Na2SO4. The solvent was removed, and the**  residue purified by chromatography on silica gel (CHCl<sub>3</sub>/MeOH/ **NH3 60:2:0.2) to give 0.28 g (21% yield) of a solid. Recrystallization from acetonitrile afforded 0.15 g of white crystals: mp 204-205 <sup>0</sup>C; IR (KBr)** *v* **3311,2996,2948,2928,2227,1612,1524, 1492,1411,1293,1236,1135 cm'<sup>1</sup> ; <sup>1</sup>H NMR (80 MHz, CDCl3)** *b*  **(TMS) 8.70 (m, 2 H, pyr), 7.87 (broad** *A, J =* **7.6 Hz, 1 H, pyr), 7.27 (m, 11 H), 6.56 (d,** *J* **= 7.6 Hz, 1 H, NH), 6.13 (d,** *J* **• 7.6 Hz, 1 H, CHPh2), 5.07 (s, 1 H, CHCN), 3.51 (m, 4 H, pip), 2.58**  (**m**, 4 H, pip). Anal.  $(C_{25}H_{25}N_5O^{-1/2}H_2O)$  C, H, N.

**l-[(2,2-Diphenylethoxy)carbonyl]-4-(3-pyridylcyanomethyl)piperazine (63). A mixture of amine 11 (1.27,6.2 mmol), 17 (2 g, 6.2 mmol), and K2CO3 (0.7 g) in DMF (15 mL) was heated at 80 "C for 12 h. The solvent was removed, and the resulting mixture was partitioned between CHCl3 and 1N NaOH solution. The organic phase was separated and dried (Na2SO4) and the solvent was removed. The residue (1.85 g) was purified by chromatography on silica gel (hexane/EtOAc, 1:1) to give 0.24 g (10%) of a colorless oil: IR (film)** *v* **3022,2948,1692,1488,1422, 1242 cm"<sup>1</sup> ; <sup>1</sup>H NMR (80 MHz, CDCl3)** *b* **(TMS) 8.75 (m, 2 H, pyr), 7.82 (d,** *J* **= 7.5 Hz, 1 H, pyr), 7.25 (m, 11 H, pyr, Ar), 4.84 (s, 1 H, CHCN), 4.55 (complex signal, 3 H1 CHCH2O)1 3.35 (m, 4 H, pip), 2.41 (m, 4 H, pip). The product was dissolved in EtOAc**  and converted to its hydrochloride by treatment with  $\text{HCl}_{\text{IR}}/$  $Et_2O$  solution to give a white solid (75% yield): mp 97-98  $\degree$ C. **Anal. (C26H28N4O2^HCl-V4H2O) C, H, N.** 

**l-(3,3-Diphenylpropyl)-4-(3-pyridylcyanomethyl)piperazine (64). Under the conditions used for the preparation of 11, 19 was condensed with pyridine-3-carboxaldehyde to give 64 as a colorless oil (51%): IR (film)** *v* **3021, 2939, 1586, 1488,1447, 1419 cm-<sup>1</sup> ; <sup>1</sup>H NMR (80 MHz, CDCl3)** *6* **(TMS) 8.74 (d,** *J* **- 2.1 Hz, 1 H, pyr), 8.59 (dd,** *J* **= 4.8 Hz,** *J* **= 1.4 Hz, 1 H, pyr), 7.84 (broad d,** *J =* **7.9 Hz, 1 H, pyr), 7.23 (m, 11 H, pyr, Ar), 4.84 (s, 1H, CHCN), 4.00 (m, 1H, Ph2CH), 2.40 (complex signal, 12 H). The product was dissolved in EtOAc and converted to its**  hydrochloride by treatment with  $\text{HCl}_{\text{(g)}}/\text{Et}_2\text{O}$  solution to give a **white solid (78%):** mp 115-122 °C. Anal.  $(C_{26}H_{28}N_4 \cdot 3HC1)$ **V2H2O) C, H, N.** 

**l-[[[AT-(3,3-Diphenylpropyl)amino]carbonyl]methyl]-4-** (3-pyridylcyanomethyl)piperazine (110). Under the condi**tions used for the preparation of 66, compound 11 was treated with** *[N-* **(3,3-diphenyIpropyl)carbonyl] methyl chloride 20 to give**  110 (80%). Recrystallization from  $CH_2Cl_2/Et_2O$  gave a white **solid: mp 120-121 <sup>0</sup>C; IR (KBr)** *v* **3357, 3020,2937,2821,1659, 1518,1488,1448,1429,1294 cm"<sup>1</sup> ; <sup>1</sup>H NMR (80 MHz, CDCl3)** *6*  **(TMS) 8.78 (d,** *J* **= 2 Hz, 1H, pyr), 8.63 (dd,** *J* **= 4.7 Hz,** *J* **= 1.3 Hz, 1H, pyr), 7.82 (d, of t,** *J* **- 1.9 Hz,** *J* **= 7.8 Hz, 1H, pyr), 7.27 (m, 11 H, Ar, pyr), 4.91 (s, 1 H, CHCN), 3.96 (t,** *J* **- 7.7 Hz, 1 H, Ph2CH), 3.27 (q,** *J* **= 6.6 Hz, 2 H, CH2NHCO), 2.93 (s, 2 H, COCH2N), 2.57 (m, 8 H, pip), 2.33 (q,** *J* **- 7.7 Hz, 2 H, Ph2- CHCH**<sub>2</sub>**).** Anal. (C<sub>28</sub>**H**<sub>31</sub>N<sub>6</sub>O-<sup>3</sup>/<sub>4</sub>**H**<sub>2</sub>O)C, H, N.

**l-[2-[JV-(Diphenylmethyl)amino]ethyl]-4-(4-pyridylcyanomethyl)piperazine(112). To a solution of 21 (2.Og, 8.1 mmol) and Et3N (2.25 mL) in CHCl3 (15 mL) was added amine 12 (1.3 g, 6.4 mmol). The mixture was heated at reflux for 2 days. The resulting solution was then washed with water, the organic phase was separated and dried (Na2SO4), and the solvent was removed. The residue (2.58 g) was purified by chromatography on silica** 

gel (EtOAc/MeOH, 3%) to afford 0.38Og (15%) of a colorless oil: <sup>1</sup>H NMR (80 MHz, CDCl<sub>3</sub>) δ (TMS) 8.65 (m, 2 H, pyr), 7.34 (m, 12 H, pyr, Ar), 4.84 (m, 1 H, CHCN), 4.81 (m, 1 H, CHPh<sub>2</sub>), 2.54  $(m, 13 \text{ H}, \text{NHC}H_2\text{C}H_2, \text{pip}, \text{NH})$ . The product was dissolved in  $CH<sub>2</sub>Cl<sub>2</sub>$  and converted to its hydrochloride by treatment with  $\mathrm{HCl}_{\textbf{(g)}}/\mathrm{Et}_2\mathrm{O}$  solution to give a white solid: mp 180–184 °C. Anal. (C26H29Ns-V2HCl-H2O) C, **H,** N.

**[l-(Ethoxycarbonyl)-4-piperidinylidene]-3-pyridylacetonitrile (22).** To a solution of NaOEt (0.49 g) in EtOH (30 mL) was added 3-pyridylacetonitrile (0.75 mL, 7 mmol) and l-(ethoxycarbonyl)-4-piperidone (1.21 g, 7 mmol), and the resulting mixture was heated at reflux for 1 h. It was allowed to cool and worked up with CHCl<sub>3</sub> and water. The organic phase was separated and dried  $(Na<sub>2</sub>SO<sub>4</sub>)$ , and the solvent was removed. The residue was purified by chromatography on silica gel (CHCl<sub>3</sub>/MeOH 5%) to afford  $1.03$  g  $(55\%)$  of a colorless oil: <sup>1</sup>H NMR  $(80$  MHz, CDCl<sub>3</sub>)  $\delta$  (TMS) 8.55 (m, 2 H, pyr), 7.68 (broad d,  $J$  = 7.0 Hz, 1 H, pyr), 7.45 (m, 1 **H,** pyr), 4.16 (q, *J* = 7.1 Hz, 2 H), 3.69 (t, *J* = 5.7 Hz, 2 H, pip), 3.51 (t, *J* = 5.6 Hz, 2 H, pip), 2.81 (t, *J =* 6.0 Hz, 2 H, pip), 2.44 (t, *J* = 6.0 Hz, 2 H, pip), 1.27 (t, *J* = 7.0 Hz, 3 H).

**(4-Piperidinylidene)-3-pyridylacetonitrile (23).** To a solution of 22 (1.36 g, 5 mmol) and sodium iodide (2.28 g) in acetonitrile (10 mL) was added trimethylsilyl chloride (0.97 mL). The resulting mixture was heated at  $50 °C$  for 2 days. It was allowed to cool, diluted with CHCl<sub>3</sub>, and washed with 0.1 N NaOH solution and  $10\%$  sodium thiosulfate solution. The organic phase was dried  $(Na_2SO_4)$ , and the solvents were removed. The residue  $(0.95 \text{ g})$  was purified by chromatography on silica gel  $(CHCl<sub>3</sub>/$ MeOH/NH<sub>3</sub> 60:2:0.2) to give 0.28 g  $(28\%)$  of a colorless oil: IR (film) v 3309, 2949, 2208, 1612, 1582, 1410 cm<sup>-1</sup>; <sup>1</sup>H NMR (80) MHz, CDCl3) *S* (TMS) 8.57 (m, 2 H, pyr), 7.68 (d of t, *J* = 8.0 Hz, *J* = 1.8 Hz, 1 H, pyr), 7.42 (dd, *J* = 7.3 Hz, *J* = 2.5 Hz, 1 H, pyr), 2.95 (m, 6 H, pip), 2.38 (t, *J* = 5.8 Hz, 2 H, pip), 2.17 (broad s, 1 H, NH).

**[l-(3,3-Diphenylpropionyl)-4-piperidinylidene]-3-pyridylacetonitrile** (77). Amine 23 was coupled with 3,3 diphenylpropionic acid using method A to give 77 (75%) as a white solid: mp 52-56 <sup>0</sup>C; IR **(KBr)** *v* 3021, 2209, 1632, 1580, 1488, 1446, 1268 cm<sup>-1</sup>; <sup>1</sup>H NMR (80 MHz, CDCl<sub>3</sub>) δ (TMS) 8.55 (m, 2 H, pyr), 7.58 (d of t, J = 8.0 Hz, *J* = 1.7 hz, 1 H, pyr), 7.24 (m, 11 H, pyr, Ar), 4.67 (t,  $J = 7.7$  Hz, 1 H, CHPh<sub>2</sub>), 3.58 (m, 4 H, pip), 3.10 (m, 2 H, CH2CON), 2.40 (m, 4 H, pip). Anal.  $(C_{27}H_{25}N_3O^{.1}/4H_2O)$  C, H, N.

**l-(Ethoxycarbonyl)-4-(3-pyridylcyanomethyl)piperidine (25).** To a cooled  $(0 °C)$  suspension of NaH  $(1 g, 21 mmol)$ in DMF (10 mL) was added 3-pyridylacetonitrile (1.9 mL, 18 mmol) in DMF (3 mL). The suspension was stirred at room temperature for 10 min, and 24 (5.9 g, 18 mmol) dissolved in DMF (5 mL) was added. The mixture was heated at 50  $\,^{\circ}$ C for 18 h. The solvents were removed, and the residue was partitioned between  $CHCl<sub>3</sub>$  and 0.05 N NaOH solution. The organic phase was separated and dried (Na<sub>2</sub>SO<sub>4</sub>), and the solvents were removed. The residue was purified by chromatography on silica gel (hexane/ EtOAc/MeOH 10:10:1) to yield 2.5  $g(51\%)$  of a colorless oil: IR (film) *v* 2977, 2238, 1686, 1470, 1424, 1377, 1277 cm<sup>-1</sup>; <sup>1</sup>H NMR (80 MHz, CDCl3) *S* (TMS) 8.60 (m, 2 H, pyr), 7.72 (broad d, *J*   $= 7.4$  Hz, 1 H, pyr), 7.36 (m, 1 H, pyr), 4.20 (m, 2 H, pip), 4.01 (q, *J* = 7.0 Hz, 2 H, OCH2CH3), 3.84 (d, *J* = 6.6 Hz, 1H, CHCN), 2.70 (m, 2 H), 2.1-1.2 (complex signal 5 H), 1.24 (t, *J* = 7.0 Hz, 3 H, OCH2CH3); <sup>13</sup>C NMR (20.15 MHz, CDCl3) *&* (TMS) 154.75 (C), 149.21 (CH), 148.73 (CH), 134.94 (CH), 116.13 (C), 60.85  $(CH<sub>2</sub>), 42.89$  (CH<sub>2</sub>), 40.46 (CH), 29.33 (CH<sub>2</sub>), 28.33 (CH<sub>2</sub>), 14.17  $(CH<sub>3</sub>)$ .

**4-(3-Pyridylcyanomethyl)piperidine** (26) was obtained from 25, under the conditions used for converting 22 into 23, as a colorless oil in a 95% yield: IR (film) *v* 3309, 2932, 2237,1572, 1423 cm"<sup>1</sup> ; <sup>1</sup>H NMR (80 MHz, CDCl3) *&* (TMS) 8.45 (m, 2 H, pyr), 7.60 (broad d, *J* = 7.4 Hz, 1 H, pyr), 7.28 (m, 1 H, pyr), 3.62 (d, *J* = 6.5 Hz, 1 H, CHCN), 3.05 (m, 2 H, pip), 2.55 (m, 2 H, pip), 2.28 (m, 1 H, NH), 1.9-1.1 (complex signal, 5 H, pip).

**l-(Diphenylacetyl-4-(3-pyridylcyanomethyl)piperidine**  (78). Amine 26 was coupled with diphenylacetic acid using method A to give 78 (45%) as a white solid: mp 159-160 °C; IR (KBr) *v* 3050, 2920, 2241, 1625, 1491, 1447, 1429 cm<sup>-1</sup>; <sup>1</sup>H NMR (80 MHz, CD3Cl) *S* (TMS) 8.55 (m, 2 H, pyr), 7.60 (broad d, *J*  = 8.2 Hz, 1 H, pyr), 7.25 (m, 11 H, pyr, Ar), 5.16 (s, 1 H, CHPh2), 4.78 (m, 1 H, pip), 4.00 (m, 1 H, pip), 3.60 (d, *J* = 6.7 Hz, 1 H, CHCN), 2.68 (m, 2 H, pip), 2.0-1.0 (complex signal, 5 H). Anal.  $(C_{26}H_{25}N_3O)$  C, H, N;

The procedure was repeated with  $[N-(diphenylmethyl)$ amino]acetic acid to give **114** (Table VI).

**l-(3,3-Diphenylpropionyi)-4-piperidone (27).** To a solution of pyridine (25.4 mL) in  $CH_2Cl_2$  (300 mL), cooled to 0 °C, was added  $CrO<sub>3</sub>$  (24.37 g) in portions. The reaction was allowed to warm to room temperature, and then l-(3,3-diphenylpropionyl)- 4-hydroxypiperidine (prepared **by** acylation of 4-hydroxypiperidine using method A)  $(12 g, 39 mmol)$  in  $CH<sub>2</sub>Cl<sub>2</sub> (100 mL)$  was added dropwise. The resulting solution was stirred at room temperature for 18 h and passed through a short column of MgSO4 which was then washed with CHCl3. After evaporation of the solvents, the crude was purified by chromatography on silica gel  $(CHCl<sub>3</sub>/MeOH 5%)$  to yield 8.7 g (73%) of 27 as a colorless oil: IR (KBr) v 3019, 2899, 1710, 1632, 1445, 1422, 1267 cm<sup>-1</sup>; <sup>1</sup>H NMR (80 MHz, CDCl<sub>3</sub>) δ (TMS) 7.26 (s, 10 H, Ph), 4.72 (t, *J* = 7.2 Hz, 1 H; CHPh2), 3.66 (m, 4 H, pip), 3.14, (d, *J* = 7.5 Hz, 2 H, CH2CON), 2.29 **(m,** 2 **H,** pip), 2.04 (m, 2 **H,** pip).

**l-(3,3-Diphenylpropionyl)-4-(3-pyridinamine)piperidine (73).** Under the conditions used for the preparation of 34, reaction of piperidone 27 with 3-aminopyridine gave 73 (13%) as a colorless oil: <sup>1</sup>H NMR (80 MHz, CDCl3) *S* (TMS) 8.0 (m, 2 H, pyr), 7.24 (m, 11 H, pyr, Ar), 6.82 (broad d, *J* = 7.3 Hz, 1 H, pyr), 4.67 (t, *J* = 7.7 Hz, 1 H, CHPh2), 3.58 (m, 4 H, pip), 3.10  $(m, 2H, CH_2CON)$ ,  $2.40$   $(m, 4H, pip)$ . The product was dissolved in EtOAc and converted to its hydrochloride by treatment with  $HCl_{(g)}/Et_2O$  solution to give a white solid: mp 105-109 °C. Anal.  $(C_{25}\dot{H}_{27}N_3O\cdot2HCl\cdot1/4H_2O)$  C, H, N.

**l-(tert-Butoxycarbonyl)-4-cyano-4-(3-pyridylamino)piperidine (29).** To a mixture of 28 (2 g, 10 mmol) and 3-aminopyridine (0.94 g, 10 mmol) in glacial HOAc (14 mL) was added KCN (0.96 g, 18 mmol). The mixture was stirred at 60 °C for 12 h. The solvent was removed, and the residue was partitioned between CHCl<sub>3</sub> and 1 N NaOH solution. The organic phase was dried over anhydrous  $Na<sub>2</sub>SO<sub>4</sub>$ , and the solvent evaporated to dryness. Purification by chromatography on silica gel (EtOAc) yielded 0.5 g (17%) of a white soild: mp 123-125 °C; IR **(KBr)**  *v* 3256, 3033, 2968, 1680, 1579, 1476, 1428 cm<sup>-1</sup>; <sup>1</sup>H NMR (80 MHz, CDCl<sub>3</sub>) δ (TMS) 8.23 (m, 2 H, pyr), 7.30 (m, 2 H, pyr), 4.06 (m, 3 H, pip), 3.31 (m, 2 H, pip), 2.6-1.6 (complex signal, 4 H, pip, NH), 1.47 (s, 9 H, BOC). Anal.  $(C_{16}H_{22}N_4O_2)$  C, H, N.

In a similar manner, compound 76 was prepared from ketone 27.

**l-[[(Diphenylmethyl)amino]acetyl]-4-cyano-4-(3-pyridylamino)piperidine (113).** A mixture of 29 (0.48 g) in dioxane (3 mL) and  $\text{HCl}_{(g)}/\text{dioxane}$  solution (3 mL) was stirred at  $0^{\circ}$ C for 30 min. Then the solvents were removed, cooled 1 N NaOH solution (10 mL) was added, and the mixture was extracted with CHCl<sub>3</sub>. The organic phase was separated and dried over anhydrous Na2SO4, and the solvent was evaporated to dryness. The residue  $(0.41 g)$  was coupled with  $[N-(diphe$ nylmethyl)amino]acetic acid<sup>16</sup> (0.48 g, 2 mmol), 1-hydroxybenzotriazole  $(0.30 \text{ g}, 2.1 \text{ mmol})$ , and DCC  $(0.42 \text{ g}, 2.06 \text{ mmol})$  using method A. Purification by chromatography yielded  $0.18$  g (20%) of an oil that crystallized upon addition of  $Et_2O$ : mp 73-75 °C; IR (KBr) v 3303, 3022, 1641, 1581, 1477, 1240 cm<sup>-1</sup>; <sup>1</sup>H NMR (80) MHz, CDCl3) *h* (TMS) 8.21 (m, 2 H, pyr), 7.30 (m, 12 H, pyr, Ar), 4.83 (s, 1 H, CHPh<sub>2</sub>), 4.22 (m, 2 H, pip), 3.29 (m, 4 H, pip, CH<sub>2</sub>-CO), 2.6-1.6 (complex signal, 5 H, pip, NH). Anal.  $(C_{26}H_{27}N_5O$  $^{5}/_{4}H_{2}O$ ) C, H, N.

**B. Biological Methods: Inhibition of Platelet Aggregation in Vitro.** Platelet-aggregation studies were done by the method of Born.<sup>19</sup> Blood was collected in 3.16% sodium citrate (1 volume for 9 volumes of blood) by cardiac puncture from male New Zealand rabbits (2-2.5 kg body weight). Platelet-rich plasma  $(PRP)$  was prepared by centrifuging the blood at  $250g$  for 10 min at 4 ° C. The PRP was diluted with platelet-poor plasma obtained by further centrifuging at 300Og for 10 min. The platelet number was adjusted to  $3.\overline{5} \times 10^5$  cells/mm<sup>3</sup>. Platelet aggregation was induced by C18-PAF  $(1.5 \times 10^{-8} \text{ M})$  and measured by using a dual-channel aggregometer Chrono-log 500. Activity of the inhibitors was expressed as the  $IC_{50}$  value, i.e. the concentration required to inhibit platelet aggregatory response by 50%. The values shown in Tables I-VI were calculated by linear regression

from a single experimental curve with no less than four data points, each point being the mean of the percentage inhibition at a given concentration obtained from one to three independent experiments.

Inhibition of PAF-Induced Hypotension in Normotensive Rats. Hypotension studies were performed as described by Baranes.<sup>20</sup> Male Sprague-Dawley rats, weighing 180-220 g, were anesthetized with sodium pentobarbital (SO mg/kg, ip). Blood pressure was recorded from the left carotid artery using a Beckman pressure transducer coupled to a Beckman R611 polygraph. Right and left femoral veins were catheterized to inject PAF  $(0.5 \mu g/kg)$  or the test compound. Test compounds were administered by intravenous injection (1 mL/kg, dissolved in saline) 3 min before PAF injection. Control animals received only the vehicle. Blood pressure was monitored and percentage inhibition of PAF-induced hypotension with respect to controls was calculated. The results were expressed as  $ID_{50}$  values, i.e. the dose of test compound required to inhibit the PAF-induced hypotension by 50%. The results were calculated by linear regression from a single experimental curve with no fewer than four points, each point being the mean of the percentage inhibition at a given dose obtained from two or more independent experiments.

Inhibition of PAF-Induced Mortality in Mice.<sup>21</sup> Groups of 10 male Swiss mice weighing 22-26 g were used. 100  $\mu$ g/kg C18-PAF plus 1 mg/kg propanolol was administered through a lateral tail vein 60 min after po administration of the test compounds (20 mL/kg) or saline (control group). Animals were

observed 2 h after the PAF injection. Following this protocol we obtained a consistent mortality of 70-100 % in the control group. Percent inhibition of mortality due to the treatments in comparison with the control group was calculated. Results were given as  $ID_{50}$  values i.e. the dose required to inhibit PAF-induced mortality by 50%. The results were calculated by linear regression from a single experimental curve with no fewer than four data points.

Statistics. Statistical analyses of pharmacological data were made using a standard pharmacology program implemented on an IBM PC.<sup>25</sup>

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Supplementary Material Available: Spectral data for compounds 31,32,35,37-43,48,49,54,56-60,65,67-69,76,81, 91-99,102-106,108,109, 111, 114, and 116 (7 pages). Ordering information is given on any current masthead page.

<sup>(25)</sup> Tallarida, R. J.; Murray, R. B. Procedure 8: Graded Dose-Response. In *Manual of Pharmacologic Calculations;* Springer-Verlag: New York, 1981; pp 14-19.